



New arene *cis*-dihydrodiol metabolites from β -bromostyrenes

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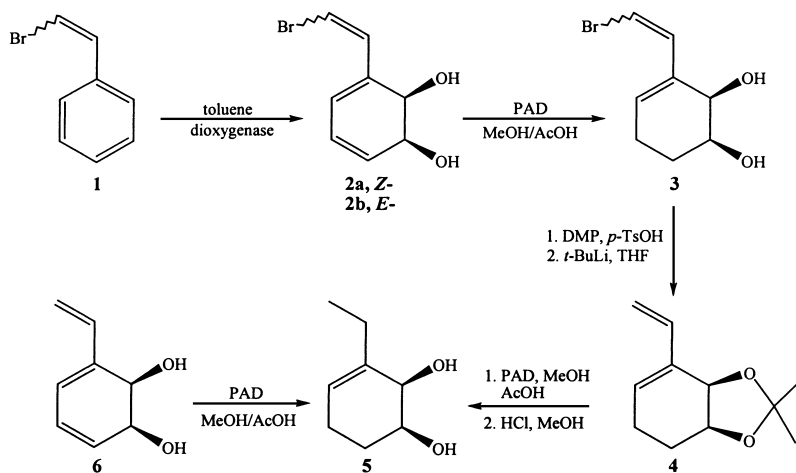
Abstract

Both (*E*)- and (*Z*)- β -bromostyrenes are substrates for toluene dioxygenase from *Escherichia coli* (JM109, pDTG601) and yield the corresponding diol metabolites in yields of 1 g/L, respectively. © 1999 Elsevier Science Ltd. All rights reserved.

The reports of new *cis*-diol metabolites¹ of aromatics continue to be published with increasing frequency as such compounds find an expanding utility in enantioselective synthesis.² Several years ago we published a report on the oxidation of *o*-, *m*-, and *p*-chlorostyrenes³ by the recombinant organism designed and provided by David Gibson.⁴ Several other bromoaromatics have also been found to be substrates for the enzyme⁵ and some of these have been and are of interest as potential starting materials for the synthesis of morphine alkaloids.⁶ Consequently, it is now possible to design a synthesis based on an *anticipated* rather than a *known* structure. In this manuscript we report the preparation and absolute stereochemistry of compounds **2a** and **2b**,⁷ which were targeted as ideal synthons for intramolecular Diels–Alder cycloadditions.

The whole cell fermentation was carried out as previously reported.⁸ Following isolation of the diols **2a** and **2b**, the absolute stereochemistry was determined as shown in Scheme 1. Diimide reduction (potassium azodicarboxylate (PAD), methanol and acetic acid, 50% yield) of the diols furnished diols **3a** and **3b**, the protection of which was followed by removal of the halogen via metal/halogen exchange (*t*-butyl lithium, 72%) to provide **4**. Further diimide reduction (PAD) of the vinyl group and deprotection afforded compound **5** (20%), which was shown to be identical (¹H NMR, melting point and [α]_D)⁹ to the material derived from styrene diol **6**¹⁰ whose absolute configuration is known.¹¹ The two compounds are excellent intermediates for an approach to morphine⁶ and progress in this area will be reported in due course.

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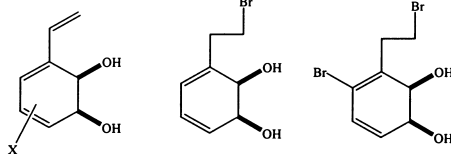
Scheme 1.

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References

- (a) Bowers, N. I.; Boyd, D. R.; Sharma, H. D.; Kennedy, M. A.; Sheldrake, G. N.; Dalton, H. *Tetrahedron: Asymmetry* **1998**, 9, 1; (b) Hudlicky, T.; Gonzalez, D.; Stabile, M.; Endoma, M. A.; Deluca, M.; Parker, D.; Gibson, D. T.; Resnick, S. M.; Whited, G. M. *J. Fluorine Chem.* **1998**, 89, 23; (c) Boyd, D. R.; Sharma, N. D.; Carroll, J. G.; Malone, J. F.; Mackerracker, D. G.; Allen, C. C. R. *J. Chem. Soc., Chem. Commun.* **1998**, 683; for other metabolites see Ref. 2a.
- (a) Hudlicky, T.; Gonzalez, D.; Gibson, D. T. *Aldrichimica Acta* **1999**, in press; (b) Brown, S. M.; Hudlicky, T. *Organic Synthesis: Theory and Applications*; Hudlicky, T., Ed.; JAI Press: Greenwich, CT, 1993; Vol. 2, p. 113; (c) Widdowson, D. A.; Ribbons, D. W.; Thomas, S. D. *Janssen Chimica Acta* **1990**, 8, 3; (d) Carless, H. A. *J. Tetrahedron: Asymmetry* **1992**, 3, 795; (e) Sheldrake, G. N. In *Chirality and Industry*; Collins, A. N.; Sheldrake, G. N.; Crosby, J., Eds.; John Wiley & Sons: 1992, p. 127; (f) Hudlicky, T.; Reed, J. R. In *Advances in Asymmetric Synthesis*; Hassner, A., Ed.; JAI Press: Greenwich, CT, 1995; p. 271; (g) Hudlicky, T. In *Green Chemistry: Designing Chemistry for the Environment*, ACS Symposium Series, Anastas, P. T.; Williamson, T., Eds.; Washington, DC, 1996; Vol. 626, p. 180; (h) Hudlicky, T.; Thorpe, A. J. *J. Chem. Soc., Chem. Commun.* **1996**, 1993; (i) Hudlicky, T. *Chem. Rev.* **1996**, 96, 3; (j) Hudlicky, T.; Entwistle, D. A.; Pitzer, K. K.; Thorpe, A. *J. Chem. Rev.* **1996**, 96, 1195; (k) Grund, A. D. *SIM News* **1995**, 45, 59; (l) Boyd, D. R. *Nat. Prod. Rep.* **1998**, 309.
- (a) Hudlicky, T.; Boros, E. E.; Boros, C. H. *Tetrahedron: Asymmetry* **1993**, 4, 1365; (b) Hudlicky, T.; Boros, E. E.; Boros, C. H. *Synlett* **1992**, 391.
- (a) Gibson, D. T.; Koch, J. R.; Kallio, R. E. *Biochemistry* **1968**, 7, 2653; (b) Gibson, D. T.; Koch, J. R.; Schuld, C. L.; Kallio, R. E. *Biochemistry* **1968**, 7, 3795; (c) Zylstra, G. J.; Gibson, D. T. *J. Biol. Chem.* **1989**, 264.
- The following styrene-related metabolites have been identified: *o*-bromostyrene: Konigsberger, K.; Hudlicky, T. *Tetrahedron: Asymmetry* **1993**, 4, 2469; Ref. 3a,b; bromoethylbenzene: Stabile, M. R.; Hudlicky, T.; Meisels, M. L. *Tetrahedron: Asymmetry* **1995**, 6, 537; *o*-bromo bromoethylbenzene: Stabile, M. R.; Hudlicky, T.; Meisels, M. L.; Butora, G.; Gum, A. G.; Fernley, S. P.; Thorpe, A. J.; Ellis, M. R. *Chirality* **1995**, 7, 556.



6. (a) Hudlicky, T.; Boros, C. H.; Boros, E. E. *Synthesis* **1992**, 174; (b) Butora, G.; Hudlicky, T.; Fernley, S. P.; Stabile, M. R.; Gum, A. G.; Gonzalez, D. *Synthesis* **1998**, 665; (c) Hudlicky, T.; Butora, G.; Gum, A. G.; Abboud, K. A. *Synthesis* **1998**, 275.
7. Experimental data for the metabolites: **2a**: (Z)- β -bromostyrene diol was isolated as a brown oil; $[\alpha]_D^{25}=40.3$ (c 1.0, CHCl₃); ¹H NMR (CDCl₃, J/Hz) δ 6.70 (d, *J*=8.3, 1H), 6.49 (d, *J*=5.6, 1H), 6.26 (d, *J*=8.3, 1H), 6.03 (ddd, *J*=9.6, 5.5, 2.0, 1H), 5.95 (dd, *J*=9.4, 2.8, 1H), 4.57 (d, *J*=6.1, 1H), 4.44–4.37 (m, 1H), 2.61 (br s, 1H), 2.31 (br s, 1H); ¹³C NMR (CDCl₃) δ 134.3, 132.2, 130.7, 127.2, 124.0, 106.0, 69.3, 67.0. **2b**: (E)- β -bromostyrene diol was isolated as a white solid, mp=48–51°C; $[\alpha]_D^{23}=83.1$ (c 0.75, CHCl₃); ¹H NMR (CDCl₃, J/Hz) δ 6.76 (d, *J*=14.0, 1H), 6.66 (d, *J*=14.2, 1H), 6.01–5.88 (m, 3H), 4.41 (td, *J*=5.9, 2.2, 1H), 4.25 (d, *J*=5.9, 1H), 2.29 (br s, 1H), 1.61 (br s, 1H); ¹³C NMR (CDCl₃) δ 136.4, 135.8, 132.2, 124.8, 124.0, 108.1, 70.0, 66.6.
8. The fermentations were carried out as described previously (Refs. 3a, 6a) in a 12 L fermentor using IPTG as the inducer. Hudlicky, T.; Stabile, M. R.; Gibson, D. T.; Whited, G. M. *Org. Synth.* **1999**, 76, 77.
9. Experimental data for **5** synthesized from β -bromostyrene diols **2a** and **2b**: mp=78–79°C; $[\alpha]_D^{25}=-156.9$ (c 1.0, CHCl₃).
10. Experimental data for **5** prepared from **6**: isolated as a white solid, mp=79–80°C; $[\alpha]_D^{27}=-150.2$ (c 1.0, CHCl₃); ¹H NMR (CDCl₃, J/Hz) δ 5.54 (br s, 1H), 3.98 (d, *J*=3.8 Hz, 1H), 3.75–3.69 (m, 1H), 2.77 (br s, 2H), 2.21–2.05 (m, 4H), 1.73–1.65 (m, 2H), 1.03 (t, *J*=7.3 Hz, 3H); ¹³C NMR (CDCl₃) δ 139.1, 123.7, 69.8, 68.8, 27.1, 25.3, 24.0, 12.4; IR (KBr/cm⁻¹) ν 3275, 1459. HRMS (EI): calcd for C₈H₁₅O₂: *m/z* 143.1776; found: 143.1072. Anal. calcd for C₈H₁₄O₂+1/3H₂O: C, 65.10; H, 9.97; found: C, 65.21; H, 9.56.
11. Hudlicky, T.; Boros, C. H. *Tetrahedron Lett.* **1993**, 34, 2557.